

## GENERAL INFORMATION

<b>Chemical Name:</b>	Calcium, Potassium, Sodium Aluminosilicate
<b>Chemical Family:</b>	Natural Zeolite
<b>Chemical Abstract Name:</b>	Clinoptilolite
<b>Chemical Formula:</b>	$(Ca, K_2, Na_2, Mg)_4 Al_8 Si_{40} O_{96} \cdot 24 H_2 O$

## MINERAL COMPOSITION \*

<b>Clinoptilolite</b>	90 - 95 %	<b>Cristobalite</b>	0 - 5 %	<b>Tridymit</b>	0 - 5 %
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\* Semi-Quantitative whole rock analysis (bulk mineralogy) has been done using powder X-ray Diffraction Method

## CHEMICAL COMPOSITION \*\*

<b>SiO<sub>2</sub></b>	65 - 72 %	<b>Fe<sub>2</sub>O<sub>3</sub></b>	0,7 - 1,9 %	<b>MnO</b>	0 - 0,08 %
<b>Al<sub>2</sub>O<sub>3</sub></b>	10 - 12 %	<b>MgO</b>	0,9 - 1,2 %	<b>Cr<sub>2</sub>O<sub>3</sub></b>	0 - 0,01 %
<b>CaO</b>	2,4 - 3,7 %	<b>Na<sub>2</sub>O</b>	0,1 - 0,5 %	<b>P<sub>2</sub>O<sub>5</sub></b>	0,02 - 0,03 %
<b>K<sub>2</sub>O</b>	2,5 - 3,8 %	<b>LOI***</b>	9 - 14 %	<b>SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub></b>	5,4 - 7,2 %

\*\* Analysed by XRF Spectrometer

\*\*\* Loss of Ignition

## PHYSICAL PROPERTIES

<b>Appearance</b>	Ivory white	<b>Oil Absorption (ml/100g)</b>	57	<b>Solubility</b>	None
<b>Smell</b>	None	<b>Abrasion (mg/100g)</b>	87	<b>pH</b>	7,0 - 8,0
<b>Porosity</b>	45 - 50 %	<b>Single Point Surface Area</b>	39 m <sup>2</sup> /g	<b>Softening Point</b>	1150 °C
<b>Hardness</b>	2 - 3 Mohs	<b>Micropore Area</b>	11 m <sup>2</sup> /g	<b>Melting Point</b>	1300 °C
<b>Mudding Down</b>	None	<b>Mesopore Area</b>	29 m <sup>2</sup> /g	<b>Bulk Density</b>	0,6 - 0,8 g / cm <sup>3</sup>
<b>Water Absorption</b>	42 - 50 %	<b>Effective Diameter of Pores</b>	4 angstrom	<b>Real Density</b>	2,2- 2,4 g / cm <sup>3</sup>
<b>Plasticity</b>	Minor				

## CATION EXCHANGE CAPACITY (CEC) †

<b>Total CEC:</b>	1,5 - 2,1 meq/g	† Methylene Blue Chloride Method
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## Major Exchangeable Cations

Rb, Li, K, Cs, NH<sub>4</sub>, Na, Ca, Ag, Cd, Pb, Zn, Ba, Sr, Cu, Hg, Mg, Fe, Co, Al, Cr.

(selectivity of above cations is a function of hydrated molecular size and relative concentrations).

## Selectivity

$Cs^+ > NH_4^+ > Pb^{2+} > K^+ > Na^+ > Ca^{2+} > Mg^{2+} > Ba^{2+} > Cu^{2+}, Zn^{2+}$

## Primary Adsorbing Gases

CO, CO<sub>2</sub>, SO<sub>2</sub>, H<sub>2</sub>S, NH<sub>3</sub>, HCHO, Ar, O<sub>2</sub>, N<sub>2</sub>, H<sub>2</sub>O, He, H<sub>2</sub>, Kr, Xe, CH<sub>2</sub>OH.

Information herein is accurate to the best of our knowledge, but may be subject to change without notice. Suggestions are made

without warranty or guarantee of results. Before using, user should determine the suitability of the product for its intended use and user

assumes the risk and liability in connection herewith.